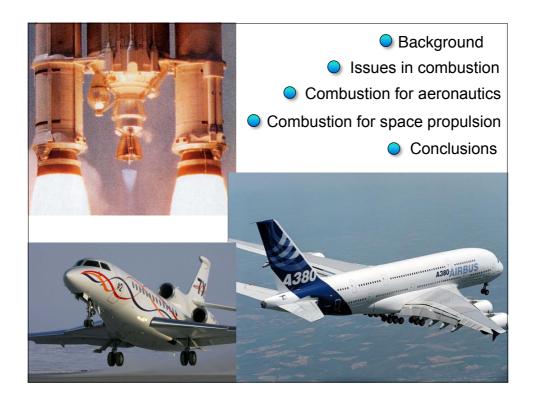


- Many thanks to Thierry Poinsot and to the scientific committee of MUSAF for inviting me to give this lecture
- It is has been a pleasure to prepare this talk and an even greater one to share it with you
- In this review, I'll underline a few of the recent developments in combustion and their application to propulsion
- To cover such a broad subject one would need a couple hundred slides and a few hours. It was heartbreaking to throw away a lot of interesting material to remain within the time allocated









Engines make the difference



Louis Blériot crosses the English channel in his Blériot XI on July 25, 1909

2013

S. Candel.



Antoinette engine 50 hp designed by Léon Levavasseur

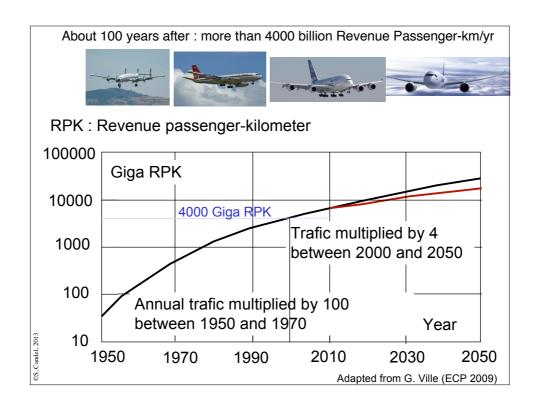


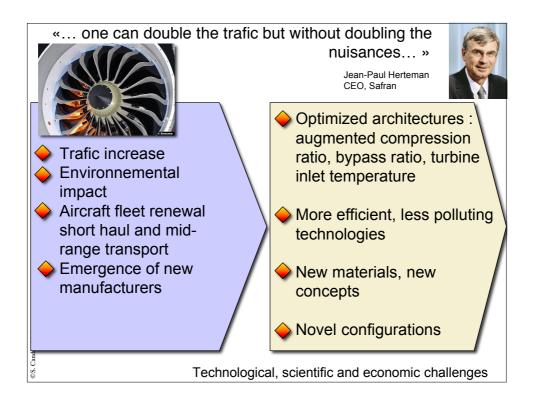
Anzani engine 3 cylinders 25 hp chosen by Blériot to cross the channel

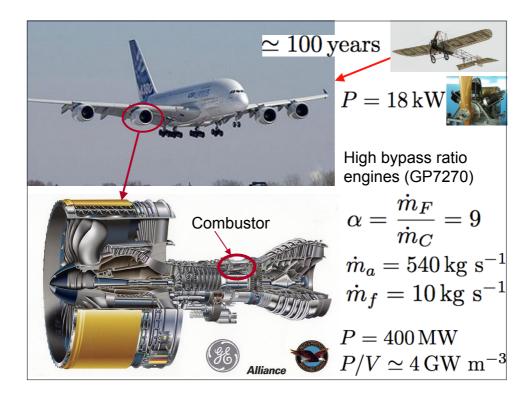
With a power per unit mass double that of the engines available elswhere, France had at the end of 1906, **the best engines in the world**, allowing the remarkable progress of its aviation in 1908 et 1909.

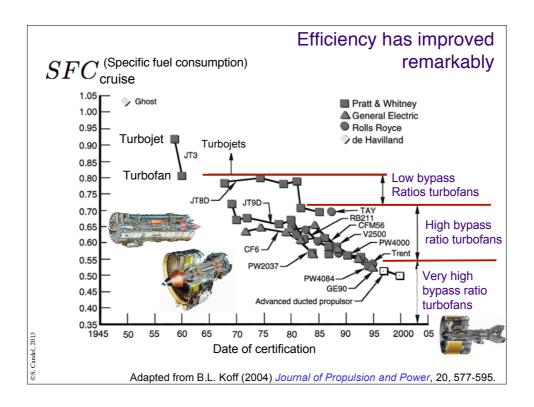
SPECIAL EDITION GENE HACKMAN THE FREE MACH COMMECTION COMMECTION

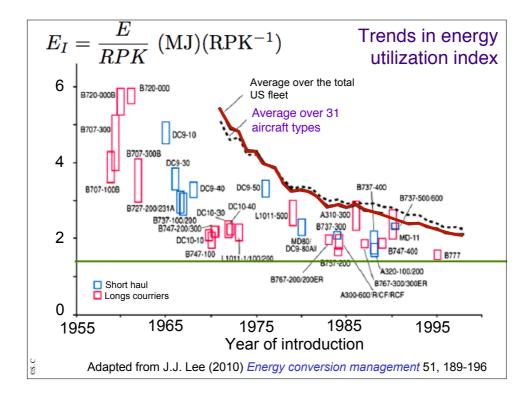
F.E.C. Culick «Aeronautics, 1898-1909 : The French American Connection » presented at *the American Society for the History of Technology*, 1987

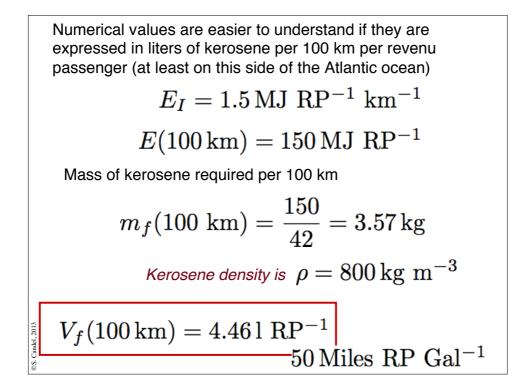


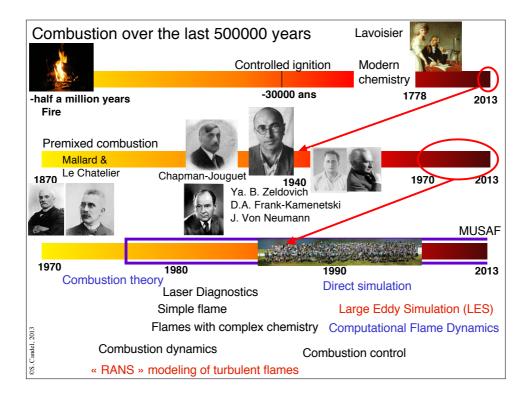




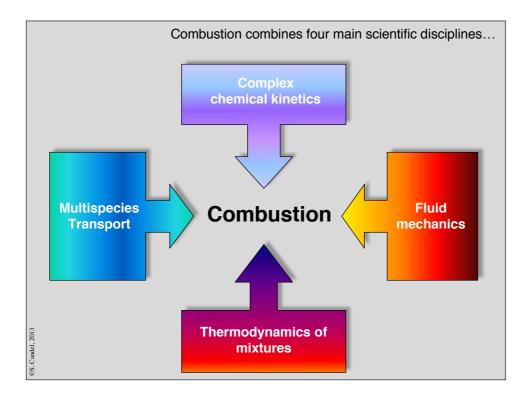


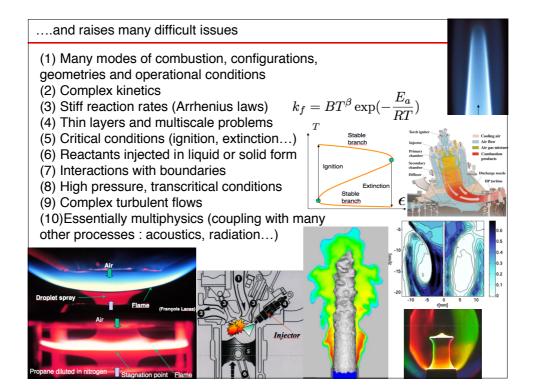








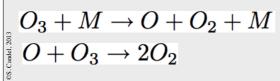




Issue 2 : Combustion involves complex chemical kinetics

Chemical conversion takes place through a large number of elementary reactions (40 species and more than 120 reactions for a propane/air flame)

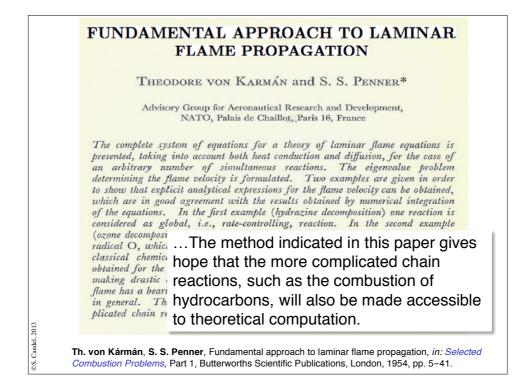
In 1954 Karman and Penner were treating the ozone flame by means of a two step reaction model





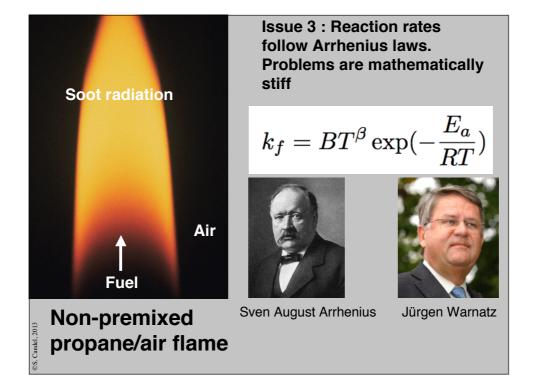
Premixed air/propane

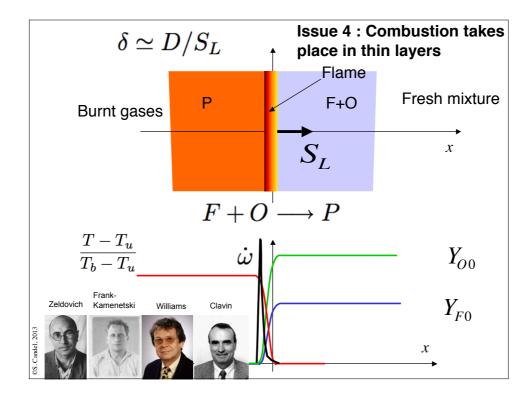


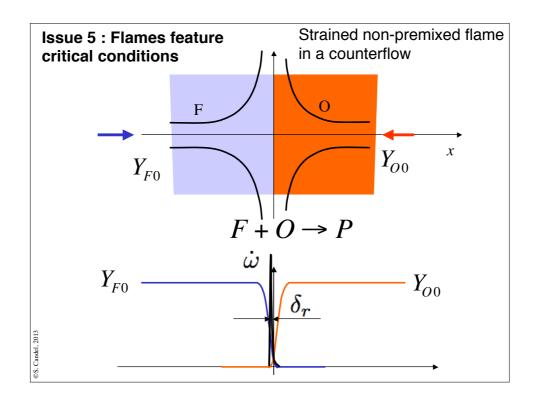


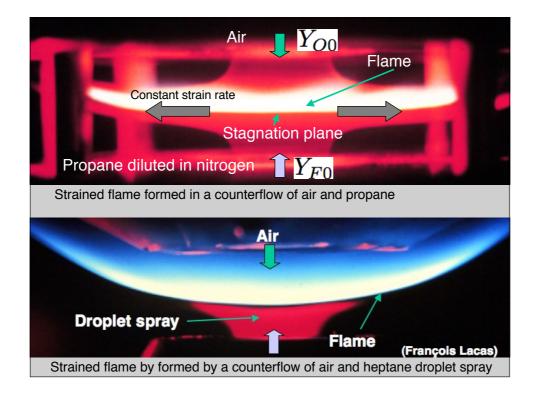
	Nb	Reaction	B	b	E
	1	$H_2 + O_2 \rightleftharpoons 2OH$	1.70E + 13	0.000	47780
	2	$OH + H_2 \rightleftharpoons H_2O + H$	1.17E + 09	1.300	3626
	3	$H + O_2 \rightleftharpoons OH + O$	2.00E+14	0.000	16800
	4	$O + H_2 \rightleftharpoons OH + H$	1.80E + 10	1.000	8826
	5^a	$H + O_2 + M \rightleftharpoons HO_2 + M$	2.10E + 18	-1.000	0.
	6	$H + O_2 + O_2 \rightleftharpoons HO_2 + O_2$	6.70E+19	-1.420	0.
	7	$H + O_2 + N_2 \rightleftharpoons HO_2 + N_2$	6.70E+19	-1.420	0.
	8	$OH + HO_2 \rightleftharpoons H_2O + O_2$	5.00E + 13	0.000	1000.
	9	$\mathrm{H} + \mathrm{HO}_2 \rightleftharpoons 2\mathrm{OH}$	2.50E + 14	0.000	1900.
	10	$O + HO_2 \rightleftharpoons O_2 + OH$	4.80E + 13	0.000	1000.
	11	$2OH \rightleftharpoons O + H_2O$	6.00E + 18	1.300	0.
	12^{b}	$H_2 + M \rightleftharpoons H + H + M$	2.23E+12	0.500	92600
	13	$O_2 + M \rightleftharpoons O + O + M$	1.85E + 11	0.500	95560
	14^{c}	$H + OH + M \rightleftharpoons H_2O + M$	7.50E + 23	-2.600	0.
	15	$H + HO_2 \rightleftharpoons H_2 + O_2$	2.50E + 13	0.000	700.
	16	$\mathrm{HO}_2 + \mathrm{HO}_2 \rightleftharpoons \mathrm{H}_2\mathrm{O}_2 + \mathrm{O}_2$	2.00E+12	0.000	0.
	17	$H_2O_2 + M \rightleftharpoons OH + OH + M$	1.30E + 17	0.000	45500
	18	$H_2O_2 + H \rightleftharpoons HO_2 + H$	1.60E + 12	0.000	3800
	19	$H_2O_2 + OH \rightleftharpoons H_2O + HO_2$	1.00E + 13	0.000	1800
Reaction mechanism for hydrogen/air oxidation $k_f = BT^b \exp(-\frac{T}{R})$ 9 species, 38 reactions					

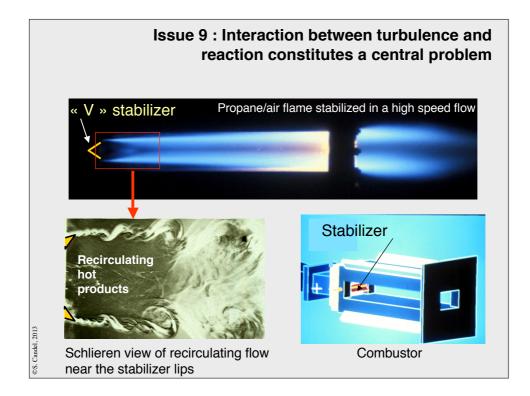
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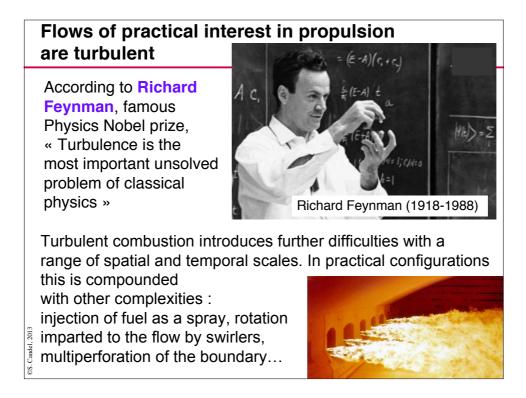


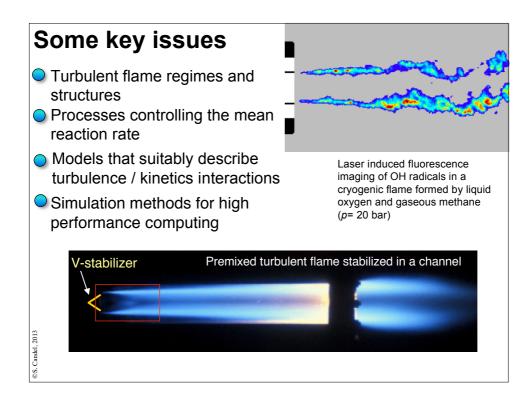


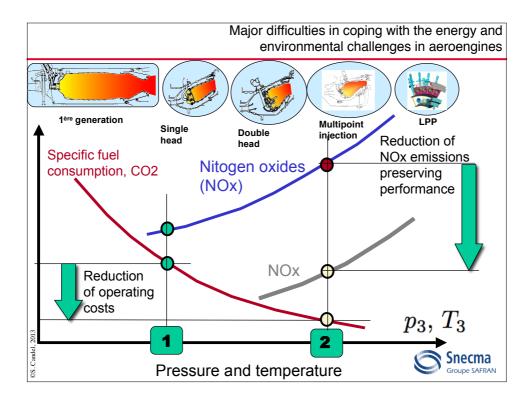




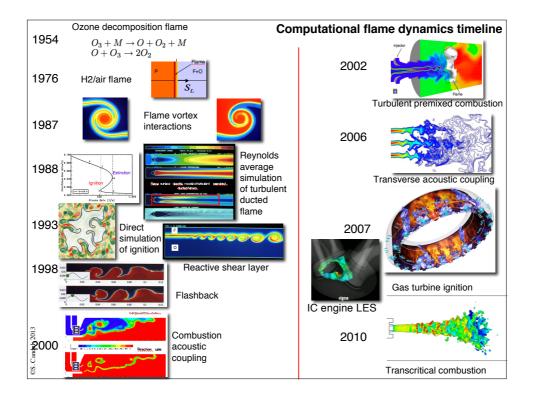


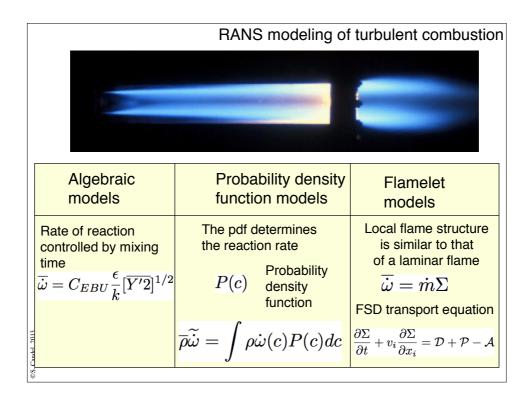


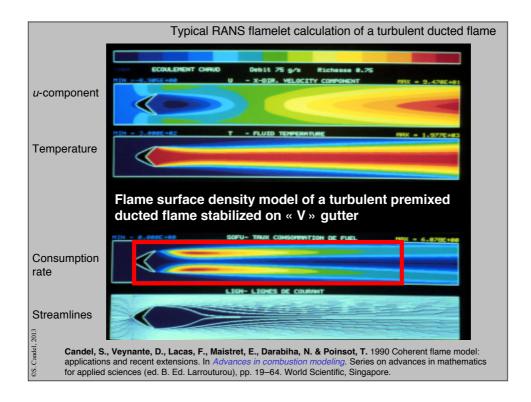


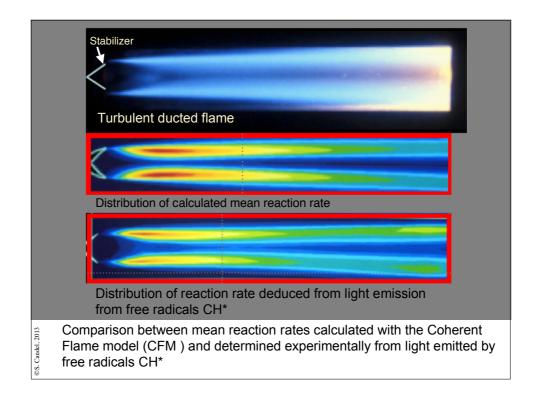


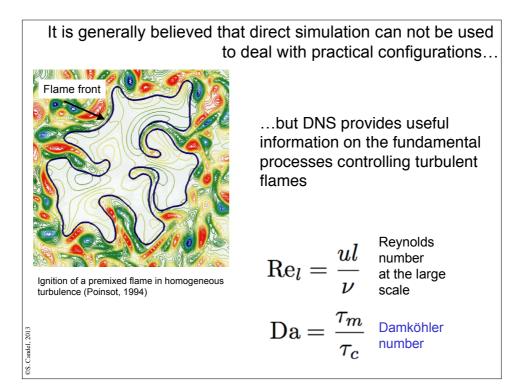
Simulation is crucial to the development of advanced combustion concepts							
Direct simulation	Large eddy simulation	Reynolds average Navier-Stokes equations					
All temporal and spatial scales are calculated	Large scales are calculated, small scales are represented by subgrid scale models	Equations are averaged and Reynolds stresses and turbulent fluxes are modeled					
Laboratory		Applications					

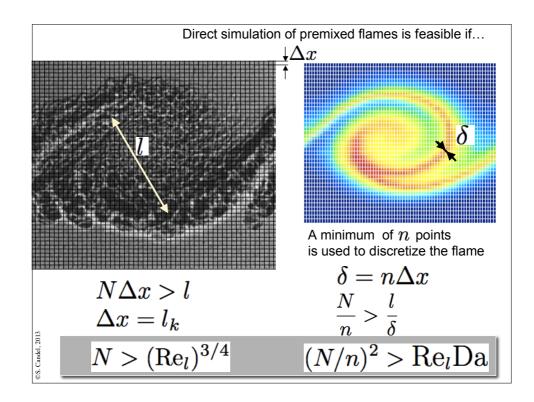


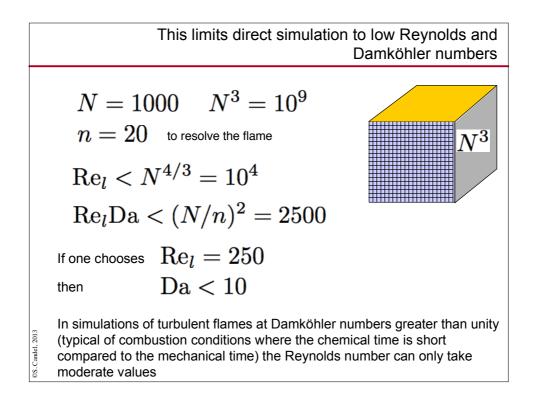


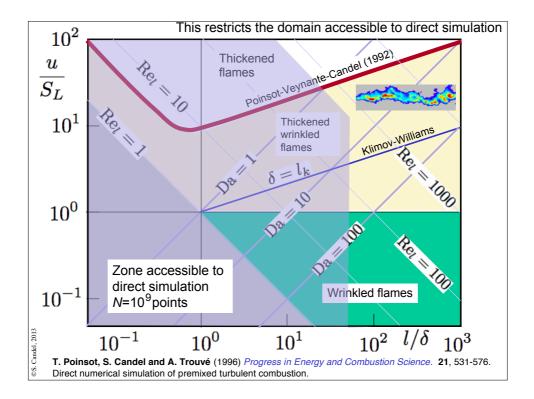


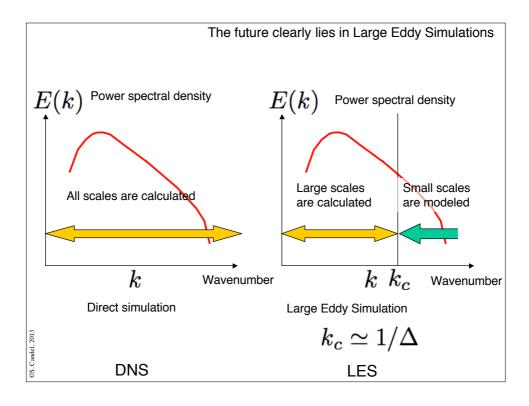


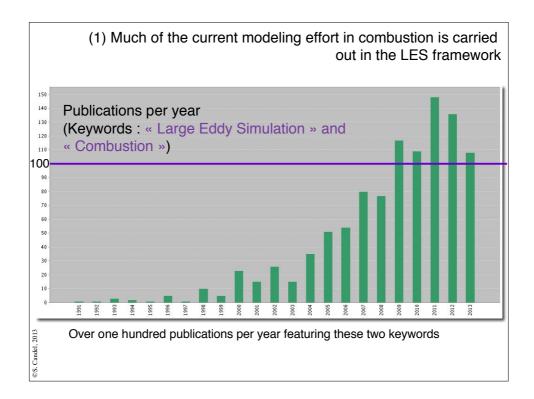


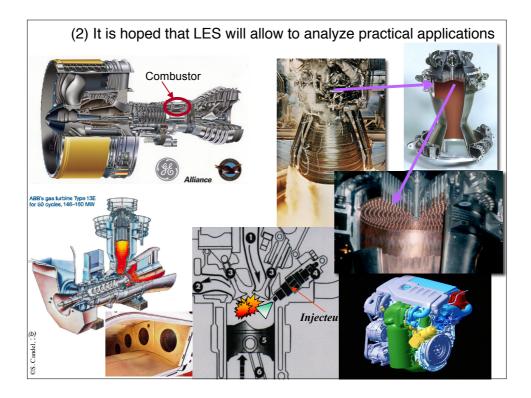


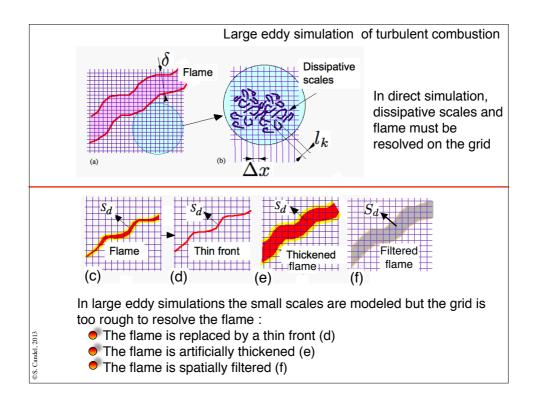


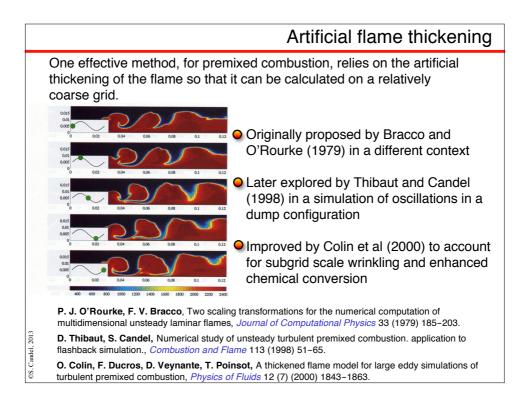


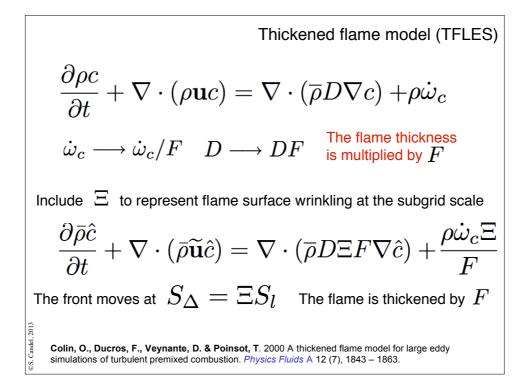


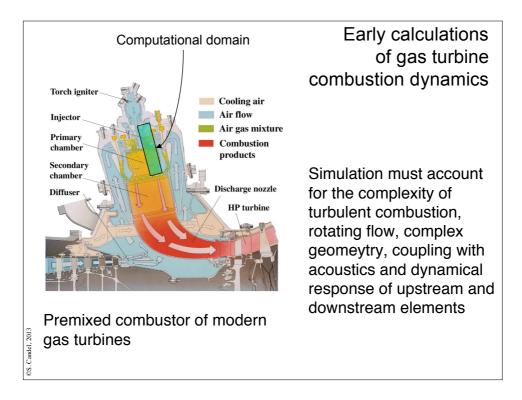


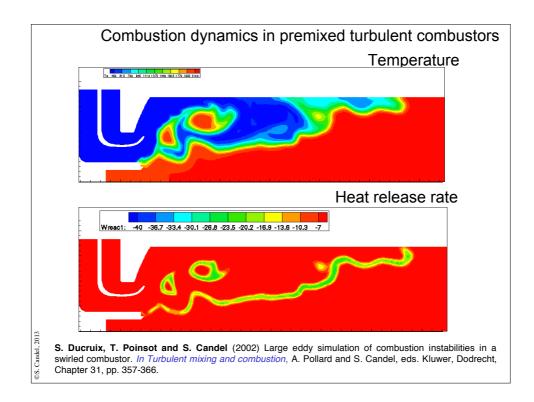


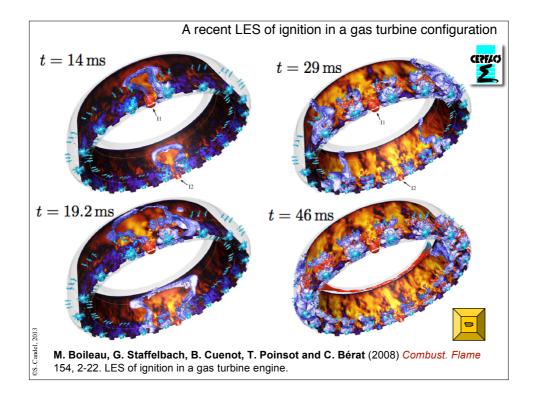


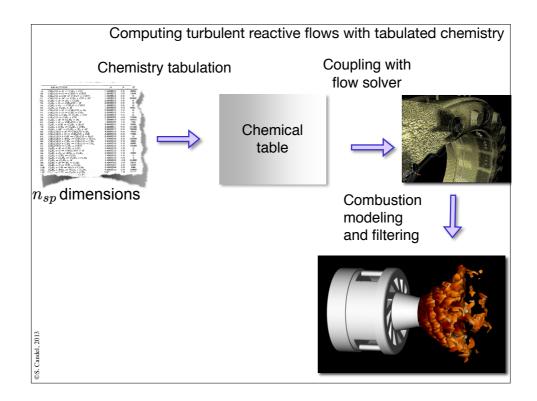


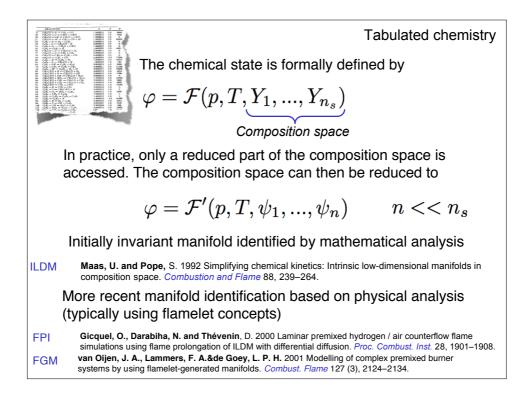


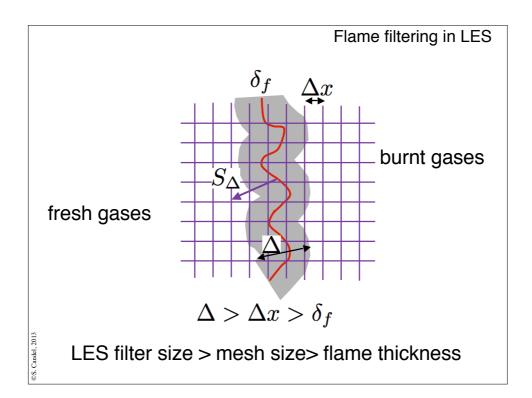


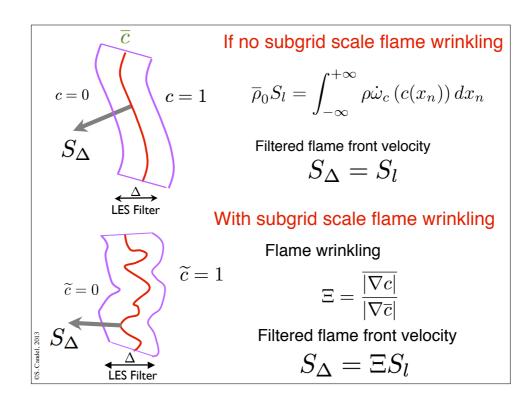


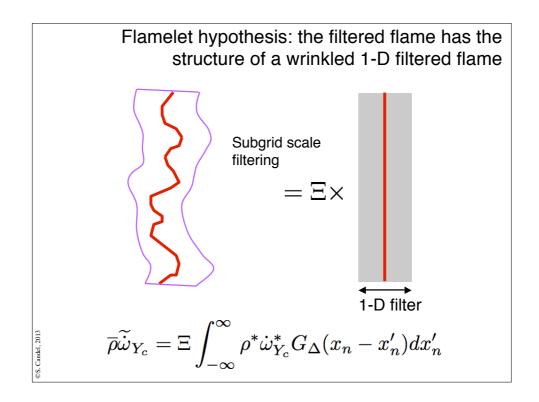


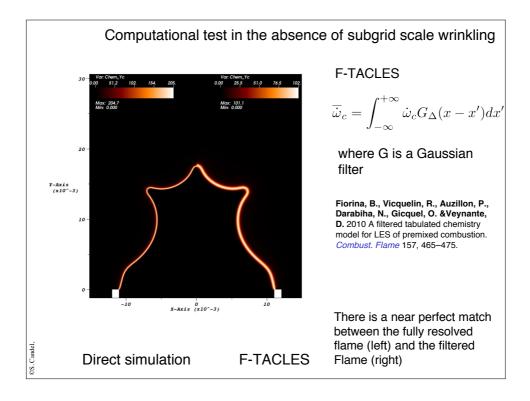


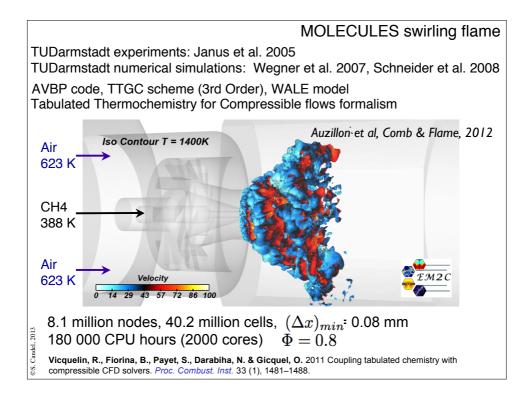


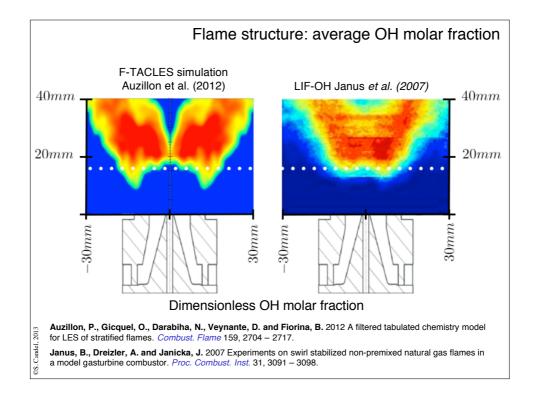


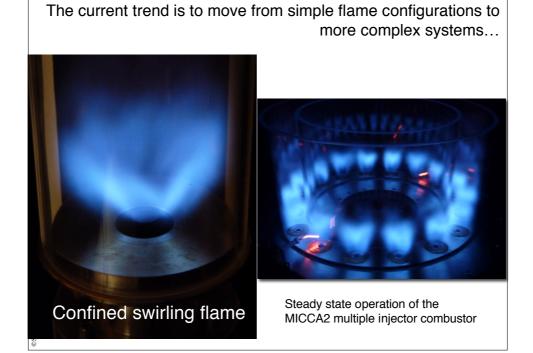


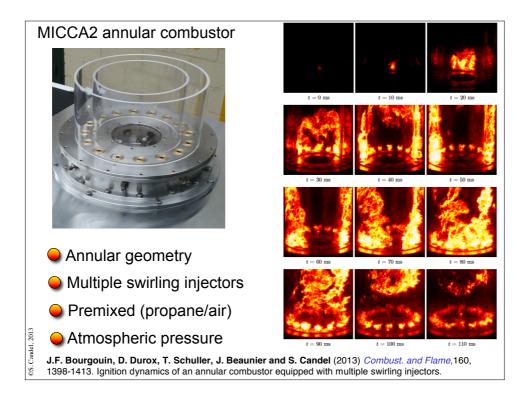


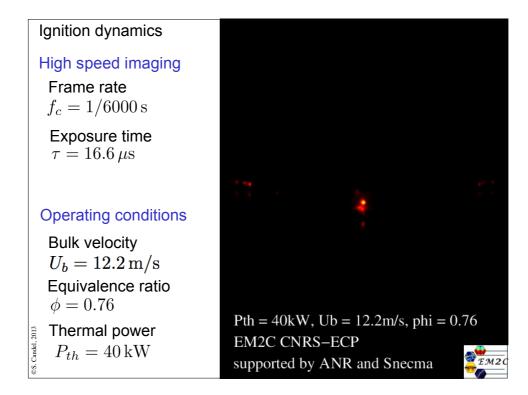


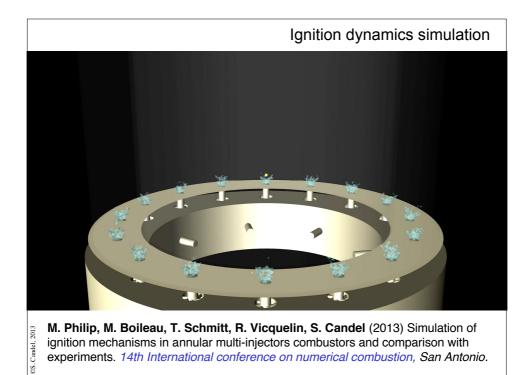


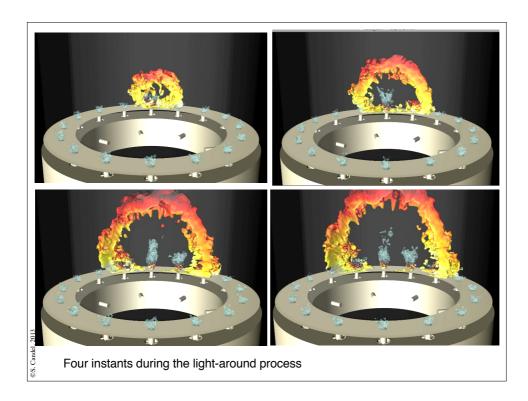


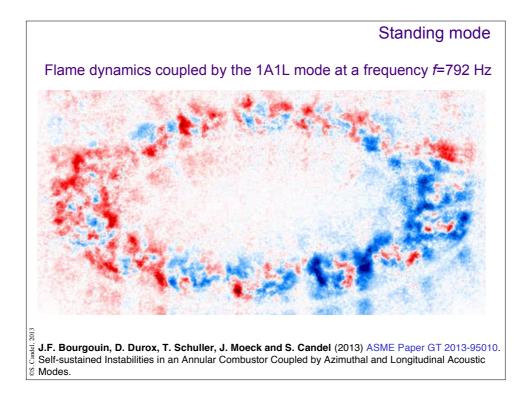


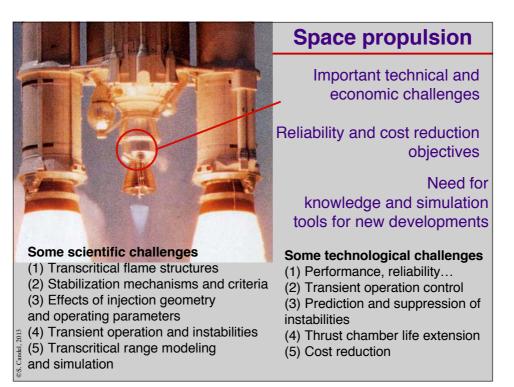


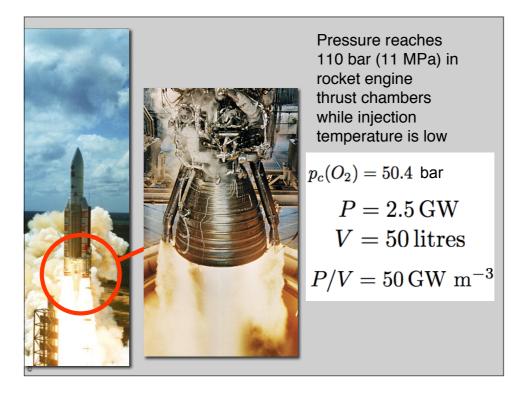


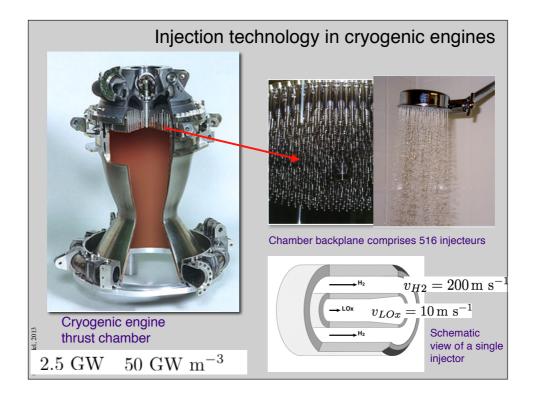


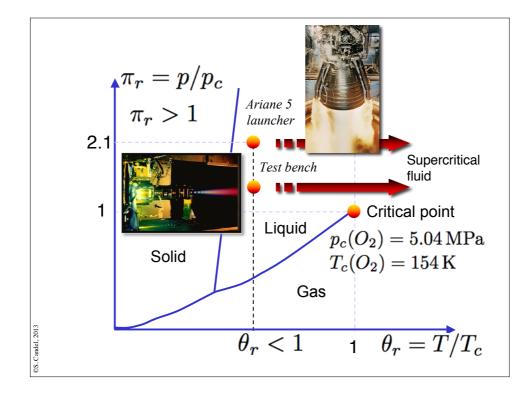












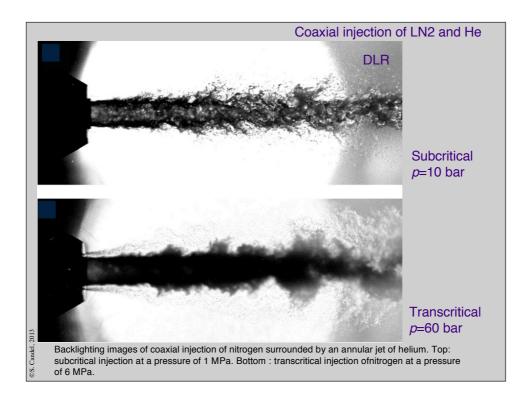


Image of the transcritical flame formed between liquid oxygen and gaseous hydrogen. A slice of the Abel transform of the average OH* emission intensity is shown in color. The average oxygen jet position is obtained by backlighting and shown on a grayscale. p= 6.3 MPa.

